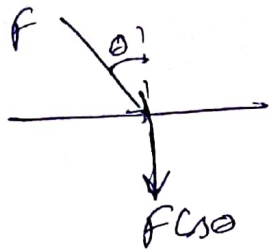


$$P = \frac{P \cos \theta}{A}$$

$$P = \frac{I A \cos^2 \theta}{A c}$$

$$P = \frac{I}{c} \cos^2 \theta$$

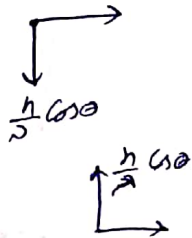


(ii) $a=0, \quad r=1$

$$P_r = \frac{h}{\lambda} (\downarrow)$$

$$P_f = \frac{h}{\lambda} (\uparrow)$$

~~$$\Delta P = \frac{2h}{\lambda}$$~~



$$\Delta P = \frac{2h}{\lambda} \cos \theta$$

$$\eta = \frac{I A \cos \theta \cdot \lambda}{h c}$$

$$F = \frac{I A \cos \theta \cdot \lambda}{h c} \cdot \frac{2h}{\lambda} \cos \theta$$

$$F = \frac{2 I A \cos^2 \theta}{c}$$

(~~h~~ hi hai)
because change in momentum is in y-direction

$$\therefore P = \frac{F}{A}$$

$$P = \frac{2 I \cos^2 \theta}{c}$$

(iii) $r = \sigma, \quad a = 1 - \sigma$

$$\eta_r = \frac{I A \cos \theta \cdot \lambda}{h c} \cdot \sigma$$

$$\eta_a = \frac{I A \cos \theta \cdot \lambda}{h c} \cdot (1 - \sigma)$$

$$\Delta P_r = \frac{2h}{\lambda} \cos \theta$$

$$\Delta P_a = \frac{h}{\lambda}$$

$$F_r = \frac{I A \cos \theta \cdot \lambda}{h c} \cdot \sigma \cdot \frac{2h}{\lambda} \cos \theta$$

$$F_r = \frac{2 I A \cos^2 \theta \cdot \sigma}{c}$$

$$F_a = \frac{I A \cos \theta \cdot \lambda}{h c} \cdot (1 - \sigma) \cdot \frac{h}{\lambda}$$

$$F_a = \frac{I A \cos \theta \cdot (1 - \sigma)}{c}$$

$$P_a = \frac{I \cos \theta \cdot (1 - \sigma)}{c A}$$

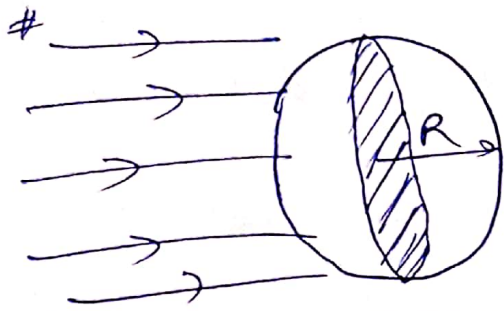
$$P_a = \frac{I \cos^2 \theta \cdot (1 - \sigma)}{c}$$

$$P_r = \frac{2 I \cos^2 \theta \cdot \sigma}{c}$$

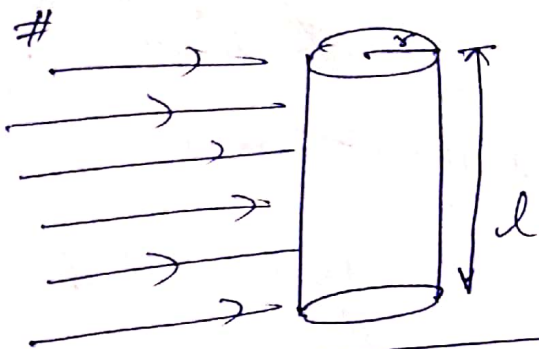
$$F_{net} = F_r + F_a$$

* The force due to radiation on a symmetric body can be given by

$$F_{net} = \frac{I}{c} \times \text{Projected Area}$$

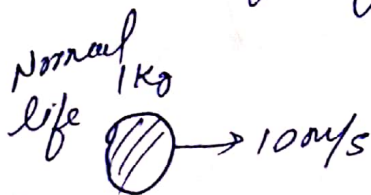
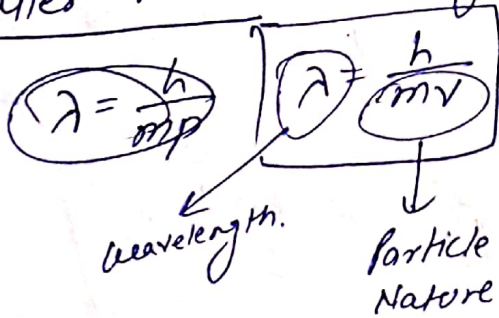


*
$$F_{net} = \frac{I}{c} \times \pi R^2$$



$$F_{net} = \frac{I}{c} \times 2\pi R l$$

Matter Waves :- de-Broglie.



$$\lambda = \frac{h}{mv}$$

$$\lambda = \frac{6.63 \times 10^{-34}}{1 \times 10}$$

$$\lambda = 6.63 \times 10^{-35} \text{ m}$$

Chhotu life body.

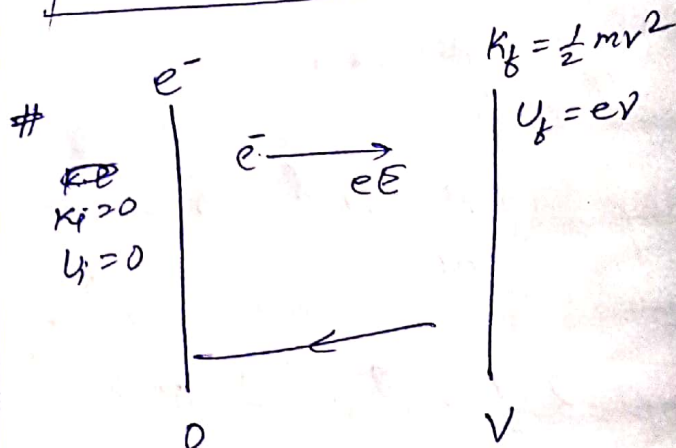
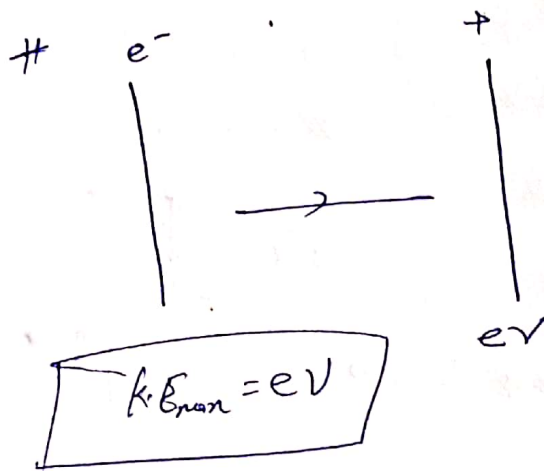


$$\lambda = \frac{h}{mv}$$

$$\lambda = \frac{6.63 \times 10^{-34}}{9.1 \times 10^{-31} \times 10}$$

$$\lambda = \frac{6.63}{9.1} \times 10^{-4} \text{ m}$$

↓
Comparable



Now from E.C

$$K_i + U_i = K_f + U_f$$

$$0 + 0 = \frac{1}{2} mv^2 - eV$$

$$eV = \frac{1}{2} mv^2$$

Now,

$$\lambda = \frac{h}{p} \quad \text{--- (1)}$$

$$\therefore \text{K.E} = \frac{p^2}{2m}$$

$$p = \sqrt{2m \cdot \text{K.E}} \quad \text{--- (2)}$$

Putting (2) in (1)

$$\lambda = \frac{h}{\sqrt{2m(\text{K.E})}}$$

$$\lambda = \frac{h}{\sqrt{2m \cdot eV}}$$

for e^-

$$\therefore \lambda = \frac{6.63 \times 10^{-34}}{\sqrt{2 \times 9.1 \times 10^{-31} \times 1.6 \times 10^{-19} \cdot V}}$$

$$\Rightarrow \lambda = \sqrt{\frac{150}{V}} \cdot \text{\AA}$$

$$\Rightarrow \lambda = \frac{12.27}{\sqrt{V}} \text{\AA}$$

(P.d.)

* for Proton,
 $m_p = 1.67 \times 10^{-27} \text{ kg}$

$$\lambda = \frac{0.286}{\sqrt{V}} \text{\AA}$$

* for deuterons.

$$m_d = 2 \times m_p$$

$$\lambda = \frac{0.202}{\sqrt{V}} \text{\AA}$$

* for α -particle :-

$$m_\alpha = 4m_p$$

$$\lambda = \frac{0.101}{\sqrt{V}} \text{\AA}$$

* find out the de-Broglie wavelength of a monoatomic gas molecule at T temp.

$$\therefore E = \frac{f}{2} nRT$$

$$* f = 3$$

$$\text{K.E} = \frac{3}{2} k \cdot T$$

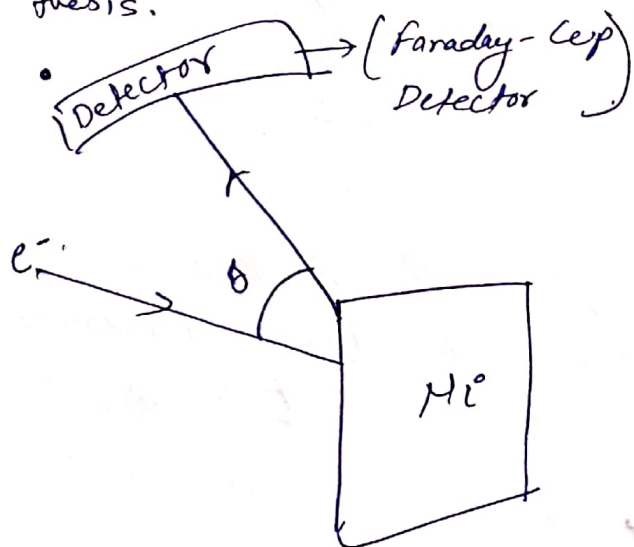
$$\lambda = \frac{h}{\sqrt{2m \cdot \frac{3}{2} kT}}$$

$$\lambda = \frac{h}{\sqrt{3mkT}}$$

→ (De-Broglie ko support).

⇒ DAVISON-GERMER Experiment :- (1924) (X-ray diffraction)

• This experiment prove that electron is a wave and supported the de-Broglie hypothesis.



• It generate e^- by an electron gun.

• These e^- are made to fall on Nickel at Normal incidence.

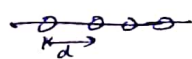
• A detector was kept at various angles and current was checked by the scattered electrons.

• Strong maxima in the intensity of reflected electron beam were found to occur at specific angles.

• Thus more current was received at some angles and these angles were analyzed.

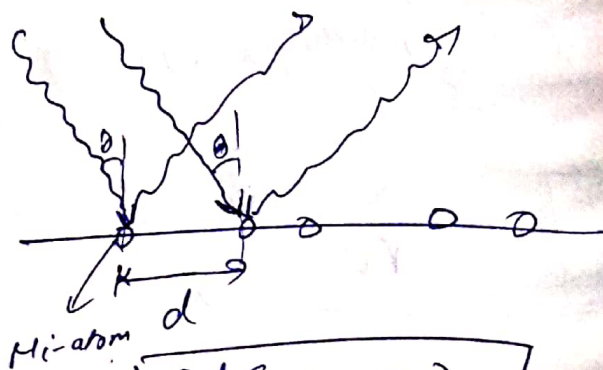
• This angular position of maxima dependent on the accelerating voltage used to produce the electron.

• This behaviour was found to be same as X-ray diffraction pattern which means that the electron beam was being diffracted. (wave hypothesis proved)



• Atoms in a surface plane are arranged in rows with a distance 'd' that can be measured by X-ray diffraction techniques.

• These rows acts like diffraction grating with distance 'd' b/w the slits.



$$2d \sin \theta = n \lambda$$